Lundurines A, B & C

Lundurines A, B and C - isolated from Kopsia tenuis, a plant native to north of Borneo

Show interesting cytotoxicity properties

These alkaloids feature a unique polyhydropyrroloazocine and cyclopropyl moiety fused to the indoline ring

Reported total syntheses of Lundurine A and B were lengthy and involve over 20 linear synthetic steps

Thus difficult to synthesize useful quantities of final targets to study broad biological assays

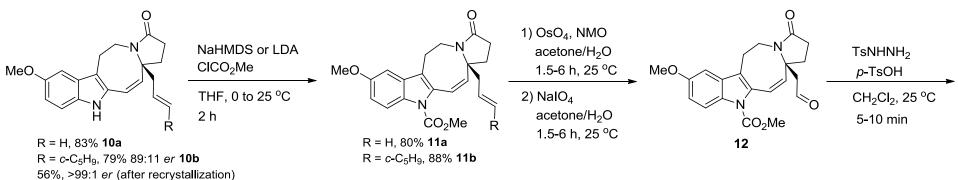
In this paper, a more efficient total synthesis (12-14 steps) of Lundurine A, B and the first total synthesis of Lundurine C (racemic and enantiopure) were reported

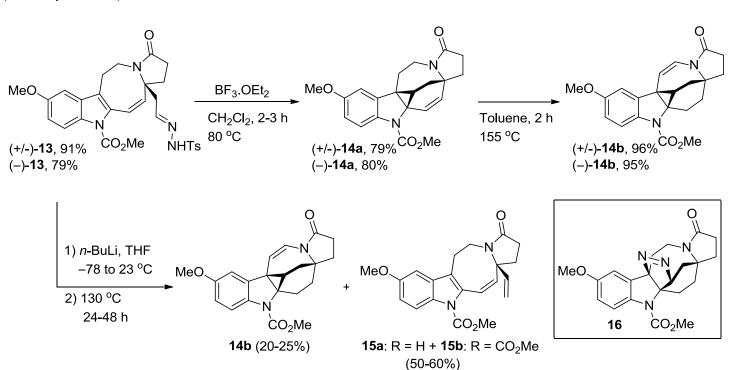
MeO
$$CO_2Me$$
 CO_2Me CO_2Me

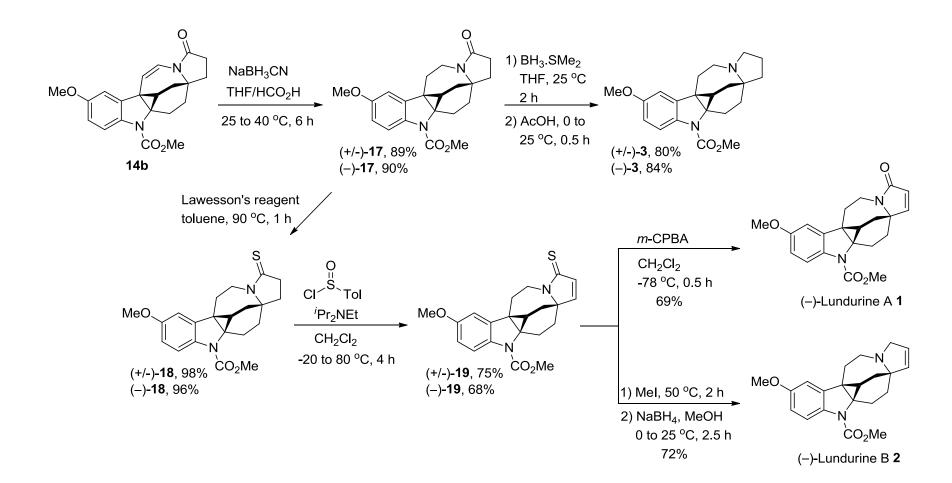
Retrosynthesis

Synthetic approach

$$\begin{array}{c} \text{MeO} \\ \text{NH}_2 \\ \text{NH}_2 \\ \text{Condensation/ Lactamization/} \\ \text{Claisen rearrangement cascade} \\ \text{Toluene/py (1:2)} \\ \text{reflux, 40 h (for 6a)} \\ \text{Toluene/Et}_3\text{N (1:2)} \\ \text{reflux, 16 h (for 6b)} \\ \text{R} \\ \text{R} \\ \text{R} \\ \text{R} \\ \text{C} \\ \text{C} \\ \text{C} \\ \text{C} \\ \text{S} \\ \text{C} \\ \text{C$$







Mechanistic explanation

Synthesis of alcohol 20

Synthesis of azo compound 21

Mechanism

 $In(OTf)_3$

0 - 25 °C

2 h, 65%

CO₂Me

 \ddot{N}_2

21

CO₂Me

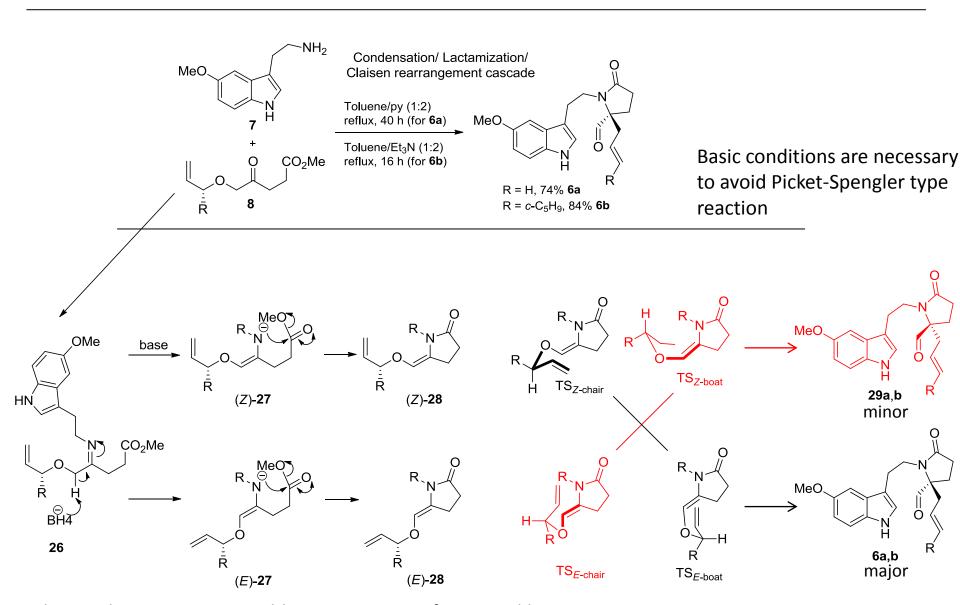
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Enzymatic Kinetic Resolution: To obtain the (S)-20

Mechanism

(S)-20

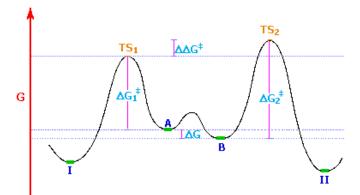
Enantioselective Claisen Rearrangement



Both Z- and E- isomers are stable even at 100 °C for several hours Curtin-Hammett principle is not applicable

Picket-Spengler type reaction: under acidic conditions

$I \xrightarrow{k_1} A \xrightarrow{K} B \xrightarrow{k_2} II$



Ohira-Bestmann reagent

MeO
$$\frac{1}{N}$$
 $\frac{1}{N}$ $\frac{1}{N}$

Curtin-Hammett principle

Mechanism

Rare 8-endo-dig gold(I) catalyzed hydroarylation

MeO AuCl (5 mol%)

$$CH_2Cl_2$$
, 25 °C

 Sh ,

 $R = H$, 88% 9a

 $R = c-C_5H_9$, 84% 9b

 $R = c-C_5H_9$, 79% 89:11 er 10b

 Sh , 79% 89:11 er (after recrystallization)

Mechanism

Overall, an 8-endo-dig reaction? Known to be catalysed by Au(III)catalysts in the presence of Ag salts No explanation has been provided why Au(I) catalyzes the reaction

Johnson-Lemiex cleavage

Mechanism

Homodienyl Retro-Ene / Ene rearrangement

MeO TsNHNH₂
$$p$$
-TsOH MeO TsNHNH₂ p -TsO

Mechanism

13b

16

$$\begin{array}{c} \text{MeO} \\ \text{MeO} \\ \text{(+/-)-17, 89\%} \\ \text{(-)-17, 90\%} \\ \text{MeO} \\ \text{Lawesson's reagent} \\ \text{Toluene} \\ \text{90 °C, 1 h} \\ \text{CO}_2\text{Me} \\ \text{19} \\ \text{MeO} \\ \text{NeO} \\ \text{S} \\ \text{S} \\ \text{NeO} \\ \text{NeO} \\ \text{NeO} \\ \text{S} \\ \text{S} \\ \text{NeO} \\ \text{NeO} \\ \text{S} \\ \text{S} \\ \text{NeO} \\ \text{NeO} \\ \text{S} \\ \text{S} \\ \text{NeO} \\ \text{$$

$$\begin{array}{c} & & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$$

MeO
$$\stackrel{\text{H}}{\longrightarrow}$$
 MeO $\stackrel{\text{N}}{\longrightarrow}$ MeO $\stackrel{\text{N}}{\longrightarrow}$ MeO $\stackrel{\text{N}}{\longrightarrow}$ MeO $\stackrel{\text{N}}{\longrightarrow}$ 19a 19a 1